

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å<sup>2</sup>)

$$U_{eq} = (1/3)\sum_i\sum_j U_{ij}a_i^*a_j^*a_i\cdot a_j$$

	x	y	z	U <sub>eq</sub>
Co	0.9268 (1)	0.4736 (1)	0.1441 (1)	0.0519 (4)
N1	0.9270 (7)	0.6305 (7)	0.1446 (3)	0.058 (2)
C2	0.9241 (9)	0.678 (1)	0.1781 (4)	0.070 (3)
C3	0.9277 (9)	0.598 (1)	0.2129 (4)	0.076 (3)
N4	0.9265 (7)	0.496 (1)	0.2010 (4)	0.065 (3)
N5	0.9261 (7)	0.3171 (7)	0.1445 (3)	0.062 (2)
C6	0.9235 (9)	0.268 (1)	0.1099 (5)	0.075 (4)
C7	0.927 (1)	0.349 (1)	0.0742 (4)	0.079 (4)
N8	0.9268 (8)	0.450 (1)	0.0876 (4)	0.068 (3)
O9	0.9226 (7)	0.6886 (8)	0.1091 (3)	0.081 (2)
O10	0.9230 (7)	0.4128 (9)	0.2277 (3)	0.081 (2)
O11	0.9196 (7)	0.2529 (7)	0.1786 (3)	0.086 (3)
O12	0.9254 (7)	0.5331 (9)	0.0608 (3)	0.082 (2)
C13	0.921 (1)	0.801 (1)	0.1841 (6)	0.107 (5)
C14	0.924 (1)	0.634 (2)	0.2549 (5)	0.107 (5)
C15	0.917 (2)	0.143 (2)	0.1028 (7)	0.133 (7)
C16	0.927 (1)	0.312 (2)	0.0325 (5)	0.110 (5)
N17	0.7262 (6)	0.3726 (6)	0.1443 (3)	0.048 (2)
C18	0.6583 (9)	0.270 (1)	0.1683 (4)	0.066 (3)
C19	0.5262 (9)	0.196 (1)	0.1671 (4)	0.068 (3)
C20	0.4547 (8)	0.2243 (8)	0.1420 (3)	0.048 (2)
C21	0.5240 (8)	0.3332 (9)	0.1183 (3)	0.055 (2)
C22	0.6536 (9)	0.4009 (9)	0.1205 (3)	0.052 (2)
C23	0.3101 (8)	0.1421 (9)	0.1411 (4)	0.062 (2)
C24	0.2555 (6)	0.1437 (8)	0.1815 (3)	0.106 (5)
C25	0.2513 (6)	0.1804 (8)	0.1060 (3)	0.104 (5)
C26	0.2747 (6)	0.0009 (8)	0.1305 (3)	0.097 (4)
C27	1.1270 (7)	0.5797 (9)	0.1464 (4)	0.077 (3)
C28	1.1876 (7)	0.6655 (9)	0.1121 (4)	0.124 (7)
C29	1.189 (1)	0.520 (1)	0.1664 (5)	0.091 (4)
C30	1.330 (1)	0.588 (1)	0.1751 (5)	0.097 (4)
C31	1.388 (2)	0.532 (2)	0.1979 (7)	0.148 (8)
C32	1.503 (2)	0.569 (2)	0.209 (1)	0.19 (1)
C33	1.578 (2)	0.546 (2)	0.225 (1)	0.26 (2)

Table 2. Selected geometric parameters (Å, °)

Co—N5	1.877 (8)	C7—C16	1.44 (2)
Co—N8	1.88 (1)	N8—O12	1.34 (2)
Co—N1	1.885 (7)	N17—C22	1.34 (1)
Co—N4	1.89 (1)	N17—C18	1.34 (1)
Co—C27	2.086 (7)	C18—C19	1.38 (1)
Co—N17	2.088 (6)	C19—C20	1.35 (1)
N1—C2	1.25 (2)	C20—C21	1.39 (1)
N1—O9	1.37 (1)	C20—C23	1.51 (1)
C2—C13	1.51 (2)	C21—C22	1.35 (1)
C2—C3	1.51 (2)	C23—C24	1.49 (2)
C3—N4	1.28 (2)	C23—C25	1.54 (1)
C3—C14	1.46 (2)	C23—C26	1.57 (1)
N4—O10	1.32 (2)	C27—C29	1.43 (1)
N5—C6	1.28 (2)	C29—C30	1.49 (2)
N5—O11	1.34 (1)	C30—C31	1.41 (2)
C6—C15	1.47 (2)	C31—C32	1.27 (2)
C6—C7	1.51 (2)	C32—C33	1.20 (3)
C7—N8	1.30 (2)		
N5—Co—C27	92.2 (4)	C27—Co—N17	177.2 (4)
N8—Co—C27	92.3 (4)	C29—C27—C28	120.9 (6)
N1—Co—C27	88.0 (3)	C29—C27—Co	116.1 (7)
N4—Co—C27	87.8 (4)	C28—C27—Co	114.9 (3)

Data reduction: REDU4 (Stoe & Cie, 1992). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: PLUTON93 (Spek, 1993). Software used to prepare material for publication: SHELXL93.

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: KA1115). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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## Tetraethylammonium (2-Thioxo-1,3-dithiole-4,5-dithiolato-S<sup>4</sup>, S<sup>5</sup>)triphenylphosphinecopper(I)

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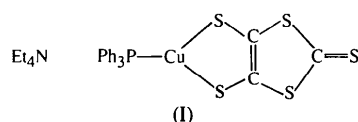
## Abstract

The structure of the title compound, (C<sub>8</sub>H<sub>20</sub>N)[Cu(C<sub>3</sub>S<sub>5</sub>)(C<sub>18</sub>H<sub>15</sub>P)], has been determined. There are two independent formula units. The Cu atom is tricoordi-

nated by two S atoms from the 2-thioxo-1,3-dithiole-4,5-dithiolate ligand and the P atom from the triphenylphosphine ligand. The bond lengths Cu—P and Cu—S are 2.181 (2) ( $\times 2$ ), and 2.238 (2) and 2.248 (2) Å, respectively.

### Comment

The chemistry of compounds involving dmit (dmit is 2-thioxo-1,3-dithiole-4,5-dithiolate) has attracted much attention in recent years (Valade *et al.*, 1991; Olk, Olk, Dietzsch, Kirmse & Hoyer, 1992) because of the good electrical conductivity of many planar bis(dmit)—metal complexes (metal = Ni, Pd, Pt, Au) (Bousseau *et al.*, 1986; Matsubayashi & Yokozawa, 1990). A series of homoleptic complexes have been obtained (Matsubayashi, Akiba & Tanaka, 1988). A new compound, (I), containing dmit and PPh<sub>3</sub> ligands is reported here.



The crystal structure of the title compound contains four formula units per unit cell and the bond lengths and angles of the two independent anions are essentially similar. Selected geometric parameters of one anion of the title compound are given in Table 2. The structure of the compound is essentially planar. In the anion, the Cu atom is coordinated in a triangular manner by two S atoms from the dmit ligand and the P atom of the PPh<sub>3</sub> ligand. The Cu—P bond length of 2.181 (2) Å is shorter than the mean Cu—P bond length found in [CuBr(Ph<sub>3</sub>P)<sub>2</sub>] [2.272 (8) Å; Davis, Beiford & Paul, 1973]. The mean Cu—S bond length of 2.237 (2) Å is shorter than that observed in both (epy)<sub>2</sub>[Cu(dmit)<sub>2</sub>] (epy is the *N*-ethylpyridinium cation) [2.271 (8) Å; Matsubayashi, Takahashi & Tanaka, 1988], in which the

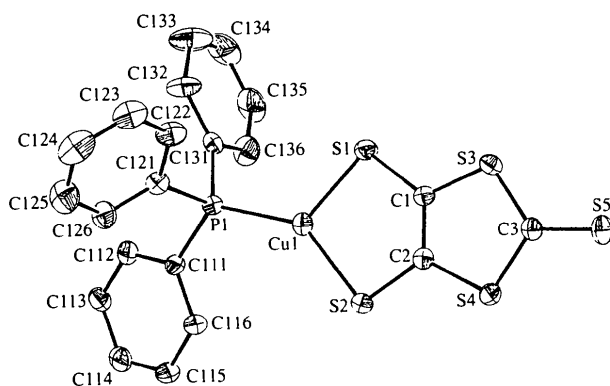


Fig. 1. ORTEP (Johnson, 1976) view of the title compound. Displacement ellipsoids are shown at the 30% probability level.

Cu atom is coordinated to two dmit ligands, and (mpy)<sub>2</sub>[Cu<sub>4</sub>(dmit)<sub>3</sub>] (mpy is the *N*-methylpyridinium cation) [2.305 (2) Å; Matsubayashi & Yokozawa, 1991].

### Experimental

The title compound was obtained from 4,5-bis(benzoylthio)-1,3-dithiol-2-thione, [Cu(PPh<sub>3</sub>)(NO<sub>3</sub>)] and Et<sub>4</sub>NBr, and crystallized from CH<sub>3</sub>OH, CH<sub>2</sub>Cl<sub>2</sub> and (C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>O.

#### Crystal data

(C<sub>8</sub>H<sub>20</sub>N)[Cu(C<sub>3</sub>S<sub>5</sub>)-  
(C<sub>18</sub>H<sub>15</sub>P)]  
*M<sub>r</sub>* = 652.44  
Triclinic  
*P* $\bar{1}$   
*a* = 13.372 (1) Å  
*b* = 15.427 (2) Å  
*c* = 16.627 (6) Å  
 $\alpha$  = 104.79 (2)°  
 $\beta$  = 77.32 (2)°  
 $\gamma$  = 104.40 (1)°  
*V* = 3167.3 Å<sup>3</sup>  
*Z* = 4  
*D<sub>x</sub>* = 1.368 Mg m<sup>-3</sup>

Mo *K*α radiation  
 $\lambda$  = 0.71073 Å  
Cell parameters from 25  
reflections  
 $\theta$  = 10.25–13.49°  
 $\mu$  = 1.08 mm<sup>-1</sup>  
*T* = 296 K  
Prism  
0.25 × 0.17 × 0.13 mm  
Red

#### Data collection

Enraf–Nonius CAD-4  
diffractometer  
 $\omega$ -2 $\theta$  scans  
Absorption correction:  
refined from  $\Delta F$   
(DIFABS; Walker &  
Stuart, 1983)  
22449 measured reflections  
11 127 independent  
reflections

6632 observed reflections  
[*I* > 3 $\sigma$ (*I*)]  
*R*<sub>int</sub> = 0.033  
 $\theta$ <sub>max</sub> = 25°  
*h* = 0 → 15  
*k* = -18 → 18  
*l* = -19 → 19  
3 standard reflections  
monitored every 300  
reflections  
intensity decay: none

#### Refinement

Refinement on *F*  
*R* = 0.069  
*wR* = 0.089  
*S* = 1.58  
6632 reflections  
667 parameters  
H atoms positioned  
geometrically (C—H =  
0.95 Å) and not refined

$w = 1/\sigma^2(F_o)$   
 $(\Delta/\sigma)_{\max} = 0.1$   
 $\Delta\rho_{\max} = 0.72 \text{ e } \text{Å}^{-3}$   
 $\Delta\rho_{\min} = -0.77 \text{ e } \text{Å}^{-3}$   
Atomic scattering factors  
from Cromer & Waber  
(1974)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å<sup>2</sup>)

$$B_{\text{eq}} = (4/3)\sum_i \sum_j \beta_{ij} a_i \cdot a_j$$

	<i>x</i>	<i>y</i>	<i>z</i>	<i>B</i> <sub>eq</sub>
Cu1	0.32873 (7)	0.25261 (6)	0.63608 (6)	3.29 (2)
Cu2	0.16205 (7)	0.74379 (6)	0.86085 (6)	3.31 (2)
S1	0.3131 (2)	0.3992 (1)	0.6731 (1)	3.52 (4)
S2	0.4770 (2)	0.2688 (1)	0.6841 (1)	3.47 (5)
S3	0.4612 (2)	0.5555 (1)	0.7624 (1)	3.30 (4)

S4	0.6048 (1)	0.4476 (1)	0.7658 (1)	3.11 (4)	S1—Cu1—S2	98.41 (6)	S1—C1—S3	119.5 (4)
S5	0.6554 (2)	0.6498 (2)	0.8340 (1)	4.37 (5)	S1—Cu1—P1	121.08 (7)	S1—C1—C2	125.5 (5)
S6	0.1753 (2)	0.5962 (1)	0.8320 (1)	3.33 (4)	S2—Cu1—P1	140.51 (7)	S3—C1—C2	114.9 (4)
S7	0.0158 (2)	0.7246 (1)	0.8089 (1)	3.25 (4)	Cu1—S1—C1	95.5 (2)	S2—C2—S4	120.0 (4)
S8	0.0307 (2)	0.4371 (1)	0.7419 (1)	3.14 (4)	Cu1—S2—C2	95.7 (3)	S2—C2—C1	124.9 (5)
S9	-0.1077 (1)	0.5441 (1)	0.7265 (1)	3.17 (4)	Cu1—P1—C111	118.5 (3)	S4—C2—C1	115.1 (4)
S10	-0.1551 (2)	0.3414 (2)	0.6602 (2)	4.67 (6)	Cu1—P1—C121	112.6 (2)	S3—C3—S4	111.5 (3)
P1	0.2129 (2)	0.1470 (1)	0.5724 (1)	2.94 (4)	Cu1—P1—C131	112.3 (3)	S3—C3—S5	124.3 (4)
P2	0.2781 (2)	0.8528 (1)	0.9209 (1)	3.17 (5)	C1—S3—C3	99.6 (3)	S4—C3—S5	124.3 (4)
N1	0.7005 (5)	0.3778 (4)	0.4499 (4)	3.1 (1)	C2—S4—C3	99.0 (3)		
N2	0.1867 (5)	0.3562 (4)	0.9507 (4)	3.9 (2)				
C1	0.4266 (5)	0.4395 (5)	0.7170 (4)	2.7 (2)				
C2	0.4946 (5)	0.3869 (5)	0.7205 (4)	2.6 (2)				
C3	0.5779 (6)	0.5566 (5)	0.7907 (4)	3.0 (2)				
C4	0.0648 (5)	0.5548 (5)	0.7841 (4)	2.6 (2)				
C5	0.0005 (5)	0.6058 (4)	0.7747 (4)	2.4 (1)				
C6	-0.0817 (6)	0.4356 (5)	0.7065 (5)	3.2 (2)				
C11	0.6267 (6)	0.3612 (6)	0.3870 (5)	4.5 (2)				
C12	0.5196 (7)	0.2978 (7)	0.4100 (7)	6.0 (3)				
C21	0.7195 (6)	0.2883 (5)	0.4608 (5)	4.0 (2)				
C22	0.7632 (9)	0.2260 (7)	0.3788 (7)	6.5 (3)				
C31	0.6571 (7)	0.4246 (6)	0.5372 (5)	4.1 (2)				
C32	0.6410 (9)	0.5221 (7)	0.5438 (7)	6.8 (3)				
C41	0.8008 (6)	0.4359 (6)	0.4106 (5)	4.2 (2)				
C42	0.8858 (7)	0.4675 (8)	0.4683 (6)	6.7 (3)				
C51	0.2895 (7)	0.3990 (6)	0.9020 (5)	5.1 (2)				
C52	0.3833 (8)	0.4140 (9)	0.9468 (6)	7.2 (3)				
C61	0.1876 (8)	0.2615 (6)	0.9616 (6)	5.7 (3)				
C62	0.203 (1)	0.1918 (7)	0.8769 (7)	8.0 (3)				
C71	0.1039 (7)	0.3545 (6)	0.8990 (5)	4.7 (2)				
C72	-0.0089 (8)	0.3119 (8)	0.9363 (7)	6.4 (3)				
C81	0.1643 (7)	0.4111 (6)	1.0406 (5)	4.6 (2)				
C82	0.1606 (9)	0.5121 (7)	1.0466 (7)	6.5 (3)				
C111	0.2515 (5)	0.0402 (5)	0.5102 (5)	2.9 (2)				
C112	0.1976 (6)	-0.0186 (5)	0.4486 (5)	3.7 (2)				
C113	0.2282 (6)	-0.0995 (5)	0.4034 (5)	3.9 (2)				
C114	0.3162 (7)	-0.1205 (5)	0.4208 (6)	4.4 (2)				
C115	0.3718 (6)	-0.0634 (6)	0.4807 (6)	4.4 (2)				
C116	0.3380 (6)	0.0193 (5)	0.5288 (5)	3.9 (2)				
C121	0.0935 (6)	0.1109 (5)	0.6456 (4)	3.3 (2)				
C122	0.0391 (7)	0.1775 (6)	0.6935 (6)	4.7 (2)				
C123	-0.0526 (7)	0.1527 (7)	0.7474 (6)	5.6 (3)				
C124	-0.0874 (8)	0.0612 (9)	0.7566 (6)	6.7 (3)				
C125	-0.0314 (9)	-0.0034 (7)	0.7105 (7)	6.4 (3)				
C126	0.0585 (7)	0.0194 (6)	0.6531 (6)	4.7 (2)				
C131	0.1682 (6)	0.1861 (5)	0.4941 (5)	3.2 (2)				
C132	0.0651 (7)	0.1983 (7)	0.5036 (6)	5.9 (2)				
C133	0.0413 (8)	0.2352 (8)	0.4405 (7)	7.9 (3)				
C134	0.1168 (9)	0.2553 (7)	0.3722 (6)	7.1 (3)				
C135	0.2171 (9)	0.2420 (6)	0.3647 (6)	6.3 (3)				
C136	0.2412 (7)	0.2084 (6)	0.4262 (6)	5.0 (2)				
C211	0.3909 (7)	0.8921 (6)	0.8438 (5)	4.2 (2)				
C212	0.3711 (9)	0.9410 (7)	0.7898 (6)	6.3 (3)				
C213	0.455 (1)	0.9710 (8)	0.7269 (7)	8.0 (4)				
C214	0.552 (1)	0.9505 (9)	0.7181 (8)	8.1 (4)				
C215	0.5699 (9)	0.9023 (9)	0.7719 (8)	7.8 (4)				
C216	0.4897 (7)	0.8710 (7)	0.8340 (6)	5.7 (3)				
C221	0.3358 (6)	0.8159 (5)	0.9935 (5)	3.4 (2)				
C222	0.3678 (7)	0.7306 (6)	0.9628 (6)	5.4 (2)				
C223	0.4097 (9)	0.6981 (7)	1.0176 (7)	7.1 (3)				
C224	0.4172 (9)	0.7466 (7)	1.0995 (7)	6.9 (3)				
C225	0.385 (1)	0.8292 (8)	1.1280 (7)	7.8 (3)				
C226	0.3449 (8)	0.8638 (7)	1.0730 (6)	6.0 (3)				
C231	0.2424 (6)	0.9588 (5)	0.9823 (5)	3.6 (2)				
C232	0.3182 (7)	1.0382 (6)	1.0033 (6)	4.9 (2)				
C233	0.2884 (8)	1.1168 (6)	1.0552 (6)	5.4 (2)				
C234	0.1811 (8)	1.1159 (6)	1.0897 (6)	5.6 (2)				
C235	0.1068 (8)	1.0392 (6)	1.0675 (6)	5.3 (2)				
C236	0.1361 (7)	0.9593 (6)	1.0149 (5)	4.4 (2)				

Table 2. Selected geometric parameters (Å, °)

Cu1—S1	2.238 (2)	S3—C1	1.744 (6)
Cu1—S2	2.231 (2)	S3—C3	1.721 (6)
Cu1—P1	2.181 (2)	S4—C2	1.750 (6)
S1—C1	1.735 (6)	S4—C3	1.736 (7)
S2—C2	1.740 (7)	S5—C3	1.648 (6)

The structure was solved by direct methods using *MITHRIL* (Gilmore, 1983). The heavy atom (Cu) was located in the *E* map and the remaining non-H atoms were located using *DIRDIF* (Beurskens, 1984). The structure was refined by a full-matrix least-squares technique with anisotropic displacement parameters for all atoms. All calculations were performed on a VAX 785 computer using the *TEXSAN* (Molecular Structure Corporation, 1987) package. The view of the molecule was produced using *ORTEPII* (Johnson, 1976).

This research was supported by State Key Structure Chemistry, Fujian Institute of Research on Structure of Matter, Chinese Academy of Sciences, and the National Science Foundation of China.

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: JZ1010). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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